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VI Semester B.Sc. Degree (CBCSS-Reg./Supple./Improv.)
Examination, April 2020
(2014 Admission Onwards)
CORE COURSE IN CHEMISTRY
6B16CHE: Physical Methods in Chemistry

Time: 3 Hours Max. Marks: 40

SECTION - A

Answer all questions. Each question carries one mark.

- 1. Give the number of fundamental vibrations for NH₃.
- 2. What is a chromophore?
- 3. Give one example for C₂ point group.
- 4. Name the internal standard in nmr.

 $(1 \times 4 = 4)$

SECTION - B. And Mad an offering 5 years a result

Answer any seven questions. Each question carries 2 marks.

- 5. Write two advantages of Raman spectra over IR spectra.
- 6. Using Woodward Feiser rule calculate λ_{max} for 3, 4 –dimethylpent-3-ene-2-one.
- 7. Sketch the nmr spectrum of acetaldehyde and identify the peaks.
- 8. State Frank condon principle.
- 9. What is McLafferty rearrangement?
- 10. How is AAS used in inorganic analysis?
- 11. What is meant by diffusion current?

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- 12. What is an inversion center?
- 13. List any two uses of nanoparticles.
- 14. What do you mean by meta stable ion?

 $(7 \times 2 = 14)$

SECTION - C

Answer any 4 questions. Each question carries 3 marks.

- 15. Explain the construction and working of dropping mercury electrode.
- 16. Discuss the chemical methods for nano particle synthesis.
- 17. What are the advantages and disadvantages of amperometric titration?
- 18. What are the factors affecting chemical shift?
- 19. Give three applications of carbon nanotubes.
- 20. Explain the terms proper and improper rotation with suitable example.

 $(3\times 4=12)$

SECTION - D

Answer any 2 questions. Each question carries 5 marks.

- 21. Discuss the methods for the characterisation of nanoparticles.
- 22. Explain the group frequency concept. What are the factors affecting IR?
- 23. Describe the theory and instrumentation of spectrophotometry.
- 24. Discuss the quantum theory of Raman scattering. (2x5=10)