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Name :			

III Semester M.Sc. Degree (CBSS – Reg./Supple./Imp.) Examination, October 2023 (2020 Admission Onwards) CHEMISTRY

CHE3C08: Inorganic Chemistry - II

Time: 3 Hours Max. Marks: 60

SECTION - A

Answer **all** questions in **one** word or **one** sentence. **Each** question carries **one** mark. (8×1=8)

- 1. Give an example of a compound which will exhibit Jahn-Teller distortion.
- 2. Tetrahedral complexes are high spin. Why?
- 3. What is the ground state term for p² configuration?
- 4. What is Neel's temperature?
- 5. Define chelate effect.
- 6. Why square planar complexes are considered as generally labile?
- 7. What is Zeise's salt? Draw its structure.
- 8. Give two examples of organometallic compounds of beryllium.

SECTION - B

Answer any eight questions. Answer may be two or three sentences. Each question carries two marks. (8×2=16)

- 9. Pt⁴⁺ ion forms octahedral whereas Pt²⁺ ion forms square planar complexes. Explain.
- 10. Draw the crystal field splitting diagram for $[CoCl_4]^{2-}$ and calculate CFSE.



- 11. Write the electronic configuration of [Fe(CN)₆]⁴⁻ on the basis of CFT and predict whether the complex is paramagnetic or diamagnetic.
- 12. Draw the Orgel diagram of transition metal complex with d² configuration.
- 13. CdCO₃ is colourless while CdS is yellow. Why?
- 14. What do you mean by spin orbit coupling in transition metal complexes?
- 15. What is trans effect? Explain using examples.
- 16. Write notes on anation reactions with suitable examples.
- 17. Explain briefly substitution reactions in square planar complexes. What are the factors influencing the substitution reactions in square planar complexes?
- 18. What are metallocenes? Draw the structure of two metallocenes.
- 19. What do you mean by migratory insertion reactions? Explain with suitable examples.
- 20. Write notes on metal alkyne complexes. Give two examples.

SECTION - C

Answer **any four** questions. Short paragraph questions. **Each** question carries **three** marks. **(4×3=12)**

- 21. Explain tetragonal distortion or Jahn-Teller distortion using suitable examples.
- 22. What is nephelauxetic effect? What is the significance of Racah parameter?
- 23. How magnetic measurements are used for the structural determination of transition metal complexes? Explain using suitable examples.
- 24. Explain thermodynamic and kinetic stability of transition metal complexes using suitable examples.
- 25. What is oxidative addition reaction? What are the characteristics of oxidative addition reactions? Discuss using examples.
- 26. Explain Monsantoacetic acid process.



SECTION - D

Answer **any four** questions. Essay type questions. **Each** question carries **six** marks. **(4×6=24)**

27. A) Explain the MO energy level diagram for octahedral, tetrahedral and square planar complexes.

OR

- B) Explain CFT. Write notes on the crystal field splitting in octahedral complexes. What are the factors affecting the magnitude of Δ ?
- 28. A) Discuss briefly Gouy method for the determination of magnetic susceptibility of complexes.

OR

- B) What are Tanabe Sugano diagrams? Draw the Tanabe Sugano diagram of d² octahedral complexes? Discuss the applications of Tanabe Sugano diagrams.
- 29. A) Explain briefly the determination of formation constants by pH metric and spectrometric methods.

OR

- B) Write notes on the redox reactions and photochemical reactions of coordination compounds.
- 30. A) Discuss the structure and bonding in ferrocene.

OR

B) Explain the mechanism of hydroformylation reaction.

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